

Communication

All-Catalytic, Efficient, and Asymmetric Synthesis of 0,0-Diheterofunctional Reduced Polypropionates via "One-Pot" Zr-Catalyzed Asymmetric Carboalumination–Pd-Catalyzed Cross-Coupling Tandem Process

Tibor Novak, Ze Tan, Bo Liang, and Ei-ichi Negishi

J. Am. Chem. Soc., **2005**, 127 (9), 2838-2839• DOI: 10.1021/ja043534z • Publication Date (Web): 11 February 2005 Downloaded from http://pubs.acs.org on March 24, 2009



One pot = 1. Zr-catalyzed asymmetric carboalumination (ZACA) 2. Pd-catalyzed vinylation

More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Links to the 19 articles that cite this article, as of the time of this article download
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

View the Full Text HTML





Published on Web 02/11/2005

All-Catalytic, Efficient, and Asymmetric Synthesis of α,*ω*-Diheterofunctional Reduced Polypropionates via "One-Pot" Zr-Catalyzed Asymmetric Carboalumination–Pd-Catalyzed Cross-Coupling Tandem Process

Tibor Novak, Ze Tan, Bo Liang, and Ei-ichi Negishi*

Herbert C. Brown Laboratories of Chemistry, Purdue University, 560 Oval Drive, West Lafayette, Indiana 47907

Received October 25, 2004; E-mail: negishi@purdue.edu

We report herein a highly efficient method for the synthesis of stereoisomerically pure ($\geq 99\%$ ee and > 50/1 dr) α,ω -diheterofunctional reduced polypropionates, the essential features of which are represented by the conversion of inexpensive (<\$1/mol) styrene into 2-methyl-4-phenyl-1-pentanol¹ (1) in 50% yield over two steps from styrene via Zr-catalyzed asymmetric carboalumination^{2,3} (ZACA reaction) and Pd-catalyzed vinylation of the in situ generated isoalkylalanes⁴ promoted by Zn(OTf)₂ (Scheme 1). Since oxidation of the first ZACA reaction with O2 gave 2-phenyl-1propanol of 89% ee by HPLC analysis of the urethane obtained by treating the alcohol with 1-a-naphthylethyl isocyanate, the diastereomeric ratio (dr) of 7.0/1 observed by 13C NMR spectroscopy for 1, before purification, indicated that the enantioface selectivity in the second ZACA reaction was 92% and that 1 should be 99% ee. The undesired diastereomers of 1 could be separated by ordinary column chromatography. For synthetic applications of 1, however, diastereomeric separation may be more efficiently and profitably performed after the conversion of the phenyl group into COOH and OH (vide infra).

To demonstrate the feasibility of using terminally phenylsubstituted methyl-branched alcohols as intermediates for the synthesis of α, ω -diheterofunctional polypropionates, a 7/1 diastereomeric mixture of 1 was acetylated first in quantitative yield and then subjected to Ru-catalyzed oxidation with NaIO4⁵ to give 5-acetoxy-2,4-dimethylpentanoic acid (2) in 75% yield over two steps. Hydrolysis of a 7/1 diastereomeric mixture of 2 with methanolic K₂CO₃ followed by acidification with 2 N HCl produced 3^6 as a 7/1 diastereomeric mixture in quantitative yield. Thus, conversion of 1 to 3 via 2 proceeded without epimerization. Comparison of the ¹H and ¹³C NMR spectral data with the reported values in conjunction with the previously established absolute stereochemistry for the first ZACA reaction^{2a,3} has firmly established that the major stereoisomers of 1-3 are the expected (2S,4S) isomers, as shown in Scheme 1. Similarly, a syn isomer of 1, i.e., (2S,4R)-1, was prepared in 47% yield over two steps as a 4.6/1 diastereomeric mixture by using (+)-(NMI)₂ZrCl₂⁷ for both ZACA reaction steps. It is noteworthy that the diastereomeric ratio is higher for the formation of the anti isomer of 1, i.e., (2S,4S)-1 (dr = 7.0/ 1), than that of the syn isomer of **1**, i.e., (2S,4R)-1 (dr = 4.6/1).⁸ In all previous cases, where 1-alkenes containing chiral alkyl groups were subjected to the ZACA reaction, there was either a minor preference for the formation of the syn isomers or essentially no internal asymmetric induction.2e-g

The high efficiency in the conversation of styrene into 1 is also critically dependent on the development of a satisfactory Pdcatalyzed vinylation of the in situ generated isoalkyldimethylalanes without oxidation to alcohols and/or conversion to iodides, the latter of which is to be subsequently lithiated with the use of 2 equiv of 'BuLi before generation of alkylzinc derivatives to be vinylated. The use of (i) $Zn(OTf)_2$ as an additive, (ii) Pd(DPEphos)Cl₂⁹ and Scheme 1^a



^{*a*} ZACA = Me₃Al (2−5 equiv), MAO (≤0.1 equiv), or H₂O (1 equiv), 3−5 mol % (+)- or (−)-(NMI)₂ZrCl₂ for (+) or (−)-ZACA, respectively, CH₂Cl₂. Pd-cat. vinylation = (a) evaporation of volatiles, (b) Zn(OTf)₂ (1− 1.5 equiv), DMF, 70 °C, (c) 3% Pd(DPEphos)Cl₂, 6% DIBAL−H, BrCH=CH₂ (5−6 equiv).

Scheme 2

(i "Hex $\sim (t)$	a) Me ₃ AI (1.5 eq) (-)-(NMI) ₂ ZrCl ₂ (3 m CH ₂ Cl ₂ ,16 h, 23 °C o) evaporation	oP%) Me nHex 75% ee 16, 23 *C Me temperature, 2 h *CH=CH4; (3-6 equiv) BCH=CH4; (3-6 equiv) *Hex *			
	additive	solvent	temp.	catalyst	yield
	(equiv)		(°C)	(%)	(%)
	ZnBr, (1)	THF	60	Pd(PPh ₂) ₄ (5)	14
	ZnBr, (1)	DMF	120	CLPd(DPEphos) (5) + 'Bu,AlH (10)	36
	ZnBr, (3)	DMF	120	Cl,Pd(DPEphos) (5) + 'Bu,AlH (10)	63
	Ze(OT0_(1)	DME	70	C1 Dd/DDEnhos) (2) + ¹ Du AILI (6)	71

ⁱBu₂AlH (DIBAL–H) in a 1:2 molar ratio as a catalyst system, and (iii) DMF as a solvent was critically important (Scheme 2). The ZACA reaction of 1-octene proceeded in 75% ee (Mosher ester analysis¹⁰ of 2-methyl-1-octanol). After Pd-catalyzed vinylation at elevated temperature (120 °C used), the product was oxidatively cleaved¹¹ to produce 3-methylnonanoic acid. Analysis by HPLC of the amide obtained by treating the acid with (*S*)-1-(α -naphthyl)ethylamine, NCP(O)(OEt)₂, Et₃N, and DMF¹² indicated the carboxylic acid to be 75% ee. Thus, no racemization took place under the conditions of the Pd-catalyzed vinylation.

Ionomycin¹³ (**4**) and borrelidin¹⁴ (**5**) are representative examples of natural products containing reduced polypropionate moieties that have been synthesized via α, ω -diheterofunctional tri- and tetramethyl-branched intermediates, respectively. Although reported syntheses appear satisfactory, all require at least in some steps the stoichiometric amounts or even excesses of enantiometrically pure starting materials and/or chiral auxiliaries. To demonstrate the feasibility of constructing some of the key intermediates employed in the previous syntheses in an "all-catalytic" manner, **6**¹³ was chosen for the synthesis of ionomycin (Scheme 3). For the synthesis of borrelidin (**5**), **7a** used by Kuwajima and Omura^{14d} and **7b** used by Theodorakis^{14c} were chosen (Scheme 4).

In the synthesis of **6** shown in Scheme 3, the most sluggish first ZACA reaction was promoted by the use of 5 equiv of Me₃Al and in situ generation of MAO by addition of 1 equiv of $H_2O.^3$ For the other cases, 2 equiv of Me₃Al were sufficient, and 0.1 equiv of MAO was used as a promoter only in the third step. Crude **8** thus obtained in 39% yield over three steps was a mixture of the desired isomer and the three major diastereomers containing minor amounts



^a See Scheme 1 for the ZACA and Pd-catalyzed vinylation conditions.

Scheme 4^a



^a See Scheme 1 for the ZACA and Pd-catalyzed vinylation conditions.

of their enantiomers in a 20/2/1.7/1 ratio by ¹³C NMR spectroscopy. After one round of chromatography (silica gel, 2/98 EtOAc—hexanes), pure **8** of dr >35/1 was obtained in 74% recovery (out of the maximum possible 81%). After acetylation in 91% yield, oxidation with NaIO₄ with cat. RuCl₃·*n*H₂O⁵ followed by reduction with BH₃·THF¹⁵ provided **9** in 55% yield over three steps or 16% yield over six steps from styrene. There was no sign of epimerization throughout these steps. Conversion of **9** into **6** (dr >50/1) in 67% yield over seven steps was achieved through the use of well-documented reactions including Dess–Martin oxidation, ole-fination with Ph₃P=CHCOOMe, and catalytic hydrogenation over 5% Rh–Al₂O₃.¹⁶

For the preparation of the borrelidin intermediates 7a and 7b (Scheme 4), all that was required to prepare the first key intermediate 10 was to repeat the ZACA-Pd-catalyzed vinylation (the second step in Scheme 3). The enantioselectivity in each of these two steps was estimated to be 90-91%. After the fourth ZACA reaction and oxidation with O_2 , 10 was obtained in 25% yield over four steps from styrene as a mixture of the desired compound and four major diastereomers containing minor amounts of their enantiomers in a 14/1.4/1.4/1.4/1 ratio. Some other very minor isomers were also detectable by ¹³C NMR spectroscopy. After one round of chromatographic purification (silica gel, 2/98 EtOAchexanes) a 22/1.6/1 mixture was obtained in 75% recovery. Evidently, two of the four asymmetric carbon centers, most probably at C2 and C4, had become essentially pure, while the C6 chiral center must have been partially purified. After acetylation, Rucatalyzed oxidation with NaIO₄ and reduction with BH₃·THF, as described above, the second key intermediate 11 was obtained in 65% yield over three steps from 10 (dr = 22/1.6/1 by ¹³C NMR). As expected, its purity was readily improved to dr >80/1 (80%

recovery) by chromatography (silica gel, 5/95 EtOAc-hexanes). Thus, pure **11** was prepared in 10% yield over seven steps and two chromatographic operations. Protection of **11** with dihydropyran and TsOH followed by ester hydrolysis with methanolic K_2CO_3 gave **7a**,^{13d} while protection of **11** with 'BuPh₂SiCl followed by deacetylation provided **7b** (99% ee by NMR analysis of the Mosher esters).^{13c}

In summary, an *efficient all-catalytic asymmetric protocol* for the synthesis of α, ω -diheterofunctional reduced polypropionates represented by **6** and **7** (dr \geq 50/1) has been developed perhaps for the first time. Although there is room for further improvements, the protocol presented herein promises to make the synthesis of α, ω -diheterofunctional reduced polypropionates more efficient and satisfactory.

Acknowledgment. This paper is dedicated to Professor Amos B. Smith on the occasion of his 60th birthday. We thank the National Institutes of Health (GM 36792), the National Science Foundation (CHE-0309613), and Purdue University for financial support. We also thank Prof. E. A. Theodorakis for spectral data of **7b**.

Supporting Information Available: Detailed experimental procedures and compound characterization data. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) Mori, I.; Bartlett, P. A.; Heathcock, C. H. J. Org. Chem. 1990, 55, 5966.
- (2) (a) Kondakov, D. Y.; Negishi, E. J. Am. Chem. Soc. 1995, 117, 10771.
 (b) Kondakov, D. Y.; Negishi, E. J. Am. Chem. Soc. 1996, 118, 1577. (c) Huo, S.; Negishi, E. Org. Lett. 2001, 3, 3253. (d) Huo, S.; Shi, J.; Negishi, E. Angew. Chem., Int. Ed. 2002, 41, 2141. (e) Negishi, E.; Tan, Z.; Liang, B.; Novak, T. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 5782. (f) Tan, Z.; Negishi, E. Angew. Chem., Int. Ed. 2004, 43, 2911. (g) Magnin-Lachaux, M.; Tan, Z.; Liang, B.; Negishi, E. Org. Lett. 2004, 6, 1425.
- (3) Wipf, P.; Ribe, S. Org. Lett. 2000, 2, 1713.
- (4) (a) Hirota, K.; Isobe, Y.; Maki, Y. J. Chem. Soc., Perkin Trans. 1 1989, 2513.
 (b) Hirota, K.; Kitade, Y.; Kanbe, Y.; Maki, Y. J. Org. Chem. 1992, 57, 5268.
- (5) Carlsen, P. H. J.; Katsuki, T.; Martin, V. S.; Sharpless, K. B. J. Org. Chem. 1981, 46, 3936.
- (6) (a) Ozegowski, R.; Kunath, A.; Schick, H. *Liebigs Ann. Chem.* 1994, 215.
 (b) Ozegowski, R.; Kunath, A.; Schick, H. *Tetrahedron: Asymmetry* 1993, 4, 695.
- (7) Erker, G.; Aulback, M.; Knickmeier, M.; Wingbermühle, D.; Krüger, C.; Nolte, M.; Werner, S. J. Am. Chem. Soc. 1993, 115, 4590.
- (8) The observed anti preference may be rationalized in terms of preferred equatorial disposition of the preexisting Me group in a putative pseudochairlike conformation arising from double π-complexation of a 14electron MeZr(NMI)₂ complex.



- (9) Kranenburg, M.; van der Burgt, Y. E. M.; Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Goubitz, K.; Fraanje, J. Organometallics 1995, 14, 3081.
- (10) Dale, J. A.; Dull, D. L.; Mosher, H. S. J. Org. Chem. 1969, 34, 2543.
- (11) Expósito, A.; Fernández-Suárez, M.; Iglesias, T.; Muñoz, L.; Riguera, R. J. Org. Chem. 2001, 66, 4206.
- (12) Bergot, B. J.; Anderson, R. J.; Schooley, D. A.; Henrick, C. A. J. Chromatogr. 1978, 155, 97.
- (13) (a) Hanessian, S.; Cooke, N. G.; DeHoff, B.; Sakito, Y. J. Am. Chem. Soc. **1990**, 112, 5276. (b) Evans, D. A.; Dow, R. L.; Shih, T. L.; Takacs, J. M.; Zahler, R. J. Am. Chem. Soc. **1990**, 112, 5290. (c) Lautens, M.; Colucci, J. T.; Hiebert, S.; Smith, N. D.; Bouchain, G. Org. Lett. **2002**, 4, 1879.
- (14) (a) Duffey, M. O.; LeTiran, A.; Morken, J. P. J. Am. Chem. Soc. 2003, 125, 1458. (b) Hanessian, S.; Yang, Y.; Giroux, S.; Mascitti, V.; Ma, J.; Raeppel, F. J. Am. Chem. Soc. 2003, 125, 13784. (c) Vong, B. G.; Kim, S. H.; Abraham, S.; Theodorakis, E. A. Angew. Chem., Int. Ed. 2004, 43, 3947. (d) Nagamitsu, T.; Takano, D.; Fukuda, T.; Otoguro, K.; Kuwajima, I.; Harigaya, Y.; Omura, S. Org. Lett. 2004, 6, 1865.
- (15) Yoon, N. M.; Pak, C. S.; Brown, H. C.; Krishnamurathy, S.; Stocky, T. P. J. Org. Chem. 1973, 38, 2786.
- (16) Takagi, Y.; Naito, T.; Nishimura, S. Bull. Chem. Soc. Jpn. 1965, 38, 2119. JA043534Z